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First Author^{*,1,3}, Second Author², and Third Author¹

¹*Department of Civil and Environmental Engineering, Carleton University, Ottawa, Canada*

²*Department of Civil Engineering, University of Ottawa, Ottawa, Canada*

³*National Research Council, Ottawa, Canada*

**Corresponding author's email: aaa@carleton.ca*

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Introduction

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Formatting Equations, Figures, and Tables

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Gibbs-Thomson Relation

The Gibbs-Thomson equation determines the critical pore radius below which the water is at a liquid state and can be described as follows:

$$T_m - T_0 = \frac{T_0 \gamma_{sl}}{\rho_i L_f R} \quad (1)$$

where T_m is the melting point of water in the pores, T_0 is the melting point of pure liquid water, γ_{sl} is the free energy coefficient of the ice–water interface, ρ_i is the ice-phase density, L_f is the latent heat of phase transformation, and R is the pore radius [1]. The variation of freezing temperature with pore radius, as predicted by Eq. 1 is shown in Fig. 1. The relation has been used for deriving the freezing characteristics curve, see for instance [2].

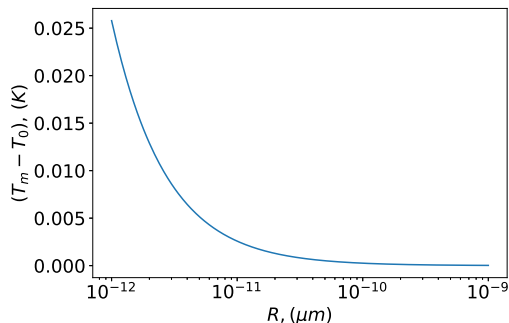


Figure 1: Gibbs-Thomson relation for freezing temperature of water in pores with varying radii.

Table 1: Parameters for water and ice for Gibbs-Thomson equation (1).

T_0 (K)	γ_{sl} (J m ⁻²)	ρ_i (kg m ⁻³)	L_f (J kg ⁻¹)
273.15	0.029	917	3.35×10^5

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References

- [1] É Devoie, S Gruber, and J McKenzie. A repository of measured soil freezing characteristic curves: 1921 to 2021. *Earth System Science Data*, 14(7):3365–3377, 2022.
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- [4] D. Evans. Unsaturated flow and transport through fractured rock-related to high-level waste repositories. Final report. Phase I. Technical report, Arizona Univ., Tucson (USA). Dept. of Hydrology and Water Resources, 1983.